

Academic Year: ( 2023 / 2024 )

Review date: 12-05-2023

Department assigned to the subject: Materials Science and Engineering and Chemical Engineering Department

Coordinating teacher: ALVAREDO OLMOS, PAULA

Type: Compulsory ECTS Credits : 6.0

Year : 2 Semester : 2

## OBJECTIVES

1. Have knowledge and understanding of the fundamentals of materials science, technology and chemistry.
2. Be aware of the multidisciplinary context of engineering.
3. Have the ability to apply their knowledge and understanding to identify, formulate, and solve materials science, technology, and chemistry problems using established methods.
4. Have the ability to design and perform experiments to solve materials science, technology, and chemistry problems, interpret data, and draw conclusions.
5. Have technical and laboratory competencies in materials science, technology and chemistry.
6. Have the ability to select and use appropriate equipment, tools, and methods to solve materials science, technology, and chemistry problems.
7. Have the ability to combine theory and practice to solve materials science, technology and chemistry problems.
8. Have an understanding of applicable methods and techniques in materials science, technology and chemistry and their limitations.

## DESCRIPTION OF CONTENTS: PROGRAMME

Introduction to Materials Science and Engineering

1. Materials Science and Engineering Framework. Concept
2. Types of Materials: Classification
3. Selection of materials
4. Relationship between structure, properties and processing

Bonding in solids

1. Types of bonding in solids
2. Bonding nature and types of materials
3. Ionic bond
4. Partial covalent bond.
5. Metallic bond. Band theory and other theories

Crystalline structures

1. Energy and crystalline lattices
2. Description of the crystalline structure
3. Main metallic structures.
4. Interstitial positions
5. Atomic positions, directions, and crystallographic planes
6. Comparison between FCC, HCP and BCC crystalline structures
7. Atomic density in crystals: linear, planar and volumetric

Defects in solids

1. Perfect and imperfect crystals: thermodynamic considerations.
2. Types of defects
3. Point defects.
4. Linear defects.
5. Planar defects.
6. Solid solutions in metals and ceramics.

Mass transport: diffusion.

1. Definition of solid state diffusion.
2. Diffusion mechanisms: substitutional and interstitial
3. Macroscopic laws governing diffusion.
4. Factors that influence the diffusion process:
5. Examples

Phase equilibrium diagrams.

1. Basic concepts. Constituents, phases and components
  2. Phase diagrams of one-component systems.
  3. Binary Systems
  4. Ceramic phase diagrams
- Charge Transport: Electrical Properties
1. Classification of materials based on their electrical properties.
  2. Basic concepts.
  3. Resistivity in Conductive Materials (metals)
  4. Semiconductor materials.
  5. Insulators and dielectric materials.

#### Mechanical properties

1. What are mechanical properties?
2. How are they evaluated? Concept of stress and strain
3. Types of mechanical tests
4. Uniaxial tensile test: nominal stress and strain.
5. Hardening mechanisms

#### Metallic Materials

1. Classification. Ferrous and non-ferrous alloys. General characteristics of:
2. Obtaining metallic materials: Solidification: Nucleation and Growth
3. Forming by plastic deformation: Strain hardening
4. Steels: Transformations in equilibrium in the Fe-C system.
5. Steels: Transformations out of equilibrium in the Fe-C system.

#### Ceramic materials

1. Classification.
2. Crystalline structures of ceramics:
3. Main ionic crystalline structures.
4. Structure of covalent ceramics.
5. Non-crystalline ceramic materials: glasses.
6. Behavior in service
7. Processing of ceramic materials

#### Polymeric materials

1. General Characteristics
2. Polymerization
3. General concepts
4. Thermal transitions:  $T_m$  and  $T_g$
5. Thermoplastics, thermosets and elastomers
6. Mechanical behavior.
7. Processing of polymer materials

#### Composite materials.

1. Classification according to the type of reinforcement and matrix
2. Type of constituents:
3. MC reinforced with particles (by dispersion and with particles)
4. Rule of mixtures
5. Fiber-reinforced MC
6. Elastic properties (MC with polymeric matrix and continuous fibers)
7. Structural materials (laminates and sandwich structures)
8. Applications and Limitations of MC
9. Processing of Composite Materials

## LEARNING ACTIVITIES AND METHODOLOGY

Masterly classes, classes to solve doubts in reduced groups, student presentations, individual tutorship and personal work of the student; oriented to acquire theoretical knowledge (3 ECTS credits).

Laboratory classes, classes for solving problems in reduced groups; individual tutorship and personal work of the student; oriented to acquire practical knowledge related to subject program (3 ECTS credits).

## ASSESSMENT SYSTEM

The assistance to the laboratory sessions is MANDATORY. The entrance to the laboratory is enabled once the student has watched the general security video and the specific video for materials lab and answered both tests correctly. THE STUDENT CAN NOT ENTER THE LABORATORY IF HE/SHE HAS NOT ANSWERED THE TESTS. THE NON-ASSISTANCE TO THE LABORATORY WITHOUT JUSTIFIED CAUSE IMPLIES SUSPENDING THE CONTINUOUS EVALUATION.

40% of the continuous assessment will be given by:

30%: Three mid-term exams, each with an calification of 10%, which will be done during class time.

10%: Lab work done outside of class time. The final note of the laboratory will be given by the behavior in the laboratory, realization of lab work and realization of a questionnaire at the end of each session.

60% of the continuous assessment will be given by the score of the final exam.

For continuous evaluation a minimum score of 4 is required on the final exam.

Normative continuous assessment:

[https://www.uc3m.es/ss/Satellite/UC3MInstitucional/en/ListadoNormativas/1371206706530/Estudios\\_de\\_Grado](https://www.uc3m.es/ss/Satellite/UC3MInstitucional/en/ListadoNormativas/1371206706530/Estudios_de_Grado)

<b>% end-of-term-examination:</b>	60
<b>% of continuous assessment (assignments, laboratory, practicals...):</b>	40

#### BASIC BIBLIOGRAPHY

- ASHBY MF, JONES DRH. Materiales para Ingeniería 1. Introducción a las propiedades, las aplicaciones y el diseño. Reverté. 2008.
- ASKELAND DR. "Ciencia e Ingeniería de los Materiales", International Thomson, 4ª Edición, Madrid, 2001.
- CALLISTER WD. "Ciencia e Ingeniería de los Materiales". Vol. I, Ed Reverté, 3ª Edición, Barcelona, 1995.
- MANGONON PL. ¿Ciencia de Materiales. Selección y Diseño. Prentice Hall, 1ª Edición, México, 2001.
- SHACKELFORD JF. "Introducción a la Ciencia de Materiales para ingenieros", Prentice Hall, 4ª Edición, Madrid, 1998.
- SMITH WF. "Fundamentos de la Ciencia e Ingeniería de Materiales", McGraw-Hill, 3ª Edición, Madrid, 2003.