

Academic Year: ( 2019 / 2020 )

Review date: 22/04/2018 16:12:12

Department assigned to the subject:

Coordinating teacher:

Type: Electives ECTS Credits : 3.0

Year : 2 Semester :

**REQUIREMENTS (SUBJECTS THAT ARE ASSUMED TO BE KNOWN)**

Introductory atomic and molecular physics.

**OBJECTIVES**

To introduce the basic computational tools to study atomic and molecular structure and dynamics with a practical point of view. Practical examples: Calculation of continuum wavefunctions; use of FFT to obtain momentum eigenfunctions; multireference calculation of potential energy surfaces for atoms and molecules; use of the R-matrix method; calculation of charge transfer (CXRS) cross sections of interest in fusion plasma diagnostic.

This course will provide the students the hand-skills computational techniques and programs to perform calculations and simulations on atomic and molecular structure and dynamics of interest in fusion plasmas

**DESCRIPTION OF CONTENTS: PROGRAMME**

- 1.- Atomic and molecular structure. Bound and continuum states. Variational, perturbational and model potential treatments. Molecular structure and symmetry. Ab-initio electronic structure of many-electron and of polyatomic molecules. Vibrational functions.
- 2.- Potential scattering and electron-atom collisions. The partial wave method. Transition probabilities. Different methods and techniques in collisions of electrons with atoms and molecules. The R-matrix method. Excitation and ionization cross sections. Resonances.
- 3.- Atomic and molecular collisions. The semiclassical approximation. Close-coupling methods for ionization and charge transfer processes. Classical CTMC methods

**LEARNING ACTIVITIES AND METHODOLOGY**

\* Teaching Methods:

Classroom lectures and classroom computational sessions. Homework assignments

\* Course Material:

Lecture notes. Computational programs. Advanced references

**ASSESSMENT SYSTEM**

<b>% end-of-term-examination/test:</b>	100
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<b>% of continuous assessment (assignments, laboratory, practicals...):</b>	0
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Evaluation shall take into account attendance and class participation, including practical classes, solution of questions, exercises and computational work to analyze an elementary system, including a theoretical introduction on one of the subjects proposed (50% of the final mark).

Written-closed book exam at the end of the semester (50% of the final mark).

**BASIC BIBLIOGRAPHY**

- B.H. Bransden and M.H.C. McDowell Charge exchange and the theory of ion-atom collisions, Clarendon, Oxford, 1992
- F. Currell The physics of multiply and highly charged ions, Kluwer Acad. , 2003
- I. N. Levine QUANTUM CHEMISTRY, Allyn and Bacon Inc. Boston, 1983
- M. Karplus and R.N. Porter ATOMS & MOLECULES, Benjamin, Menlo Park, 1970

#### ADDITIONAL BIBLIOGRAPHY

- W H Press, S.A. Tulkowsky, W. T. Vetterling, and B.P. Flannery Numerical Recipes, Cambridge University Press, 1992